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YOUR WATER HEATER IS A TICKING TIME BOMB — AND THE INSTALLER MAY HAVE PULLED THE PIN

How Direct Copper-to-Galvanized Connections Silently Destroy Water Heater Installations — And What a Proper Fix Looks Like

VENICE, FL — Every year in Southwest Florida, homeowners replace water heaters years ahead of schedule because of a problem that has nothing to do with the tank itself. The tank is fine. The issue is what was done at the connections on top of it. A licensed plumber ran copper pipe directly onto the galvanized steel nipples threaded into the heater, skipped the correct transition fittings, and walked away. That mistake set an invisible clock ticking the moment the water was turned back on.

Mike Reynolds, owner of Plumbing Detectives LLC and a licensed plumbing contractor (CFC1434139) serving Venice, Sarasota, Englewood, North Port, and Nokomis, has pulled hundreds of these installations apart. "It's one of the most common things I see," Reynolds says. "Someone installs a water heater and connects the copper pipe straight to the galvanized nipple. No brass transition. No buffer. Just copper threaded directly onto galvanized steel. By the time the homeowner notices something's wrong, the nipple is eaten through or the connection is weeping and the subfloor is wet."

This article explains exactly why that installation fails — the electrochemistry behind it, not just the rule — and what a correct installation looks like.

THE PROBLEM: COPPER ON GALVANIZED IS NOT A MINOR SHORTCUT — IT IS A GUARANTEED FAILURE

Water heaters manufactured today ship with galvanized steel nipples already installed in the hot and cold ports at the top of the tank. Galvanized means the steel was coated with zinc to protect it from rust. That coating is the first line of defense for the nipple. The problem begins the moment a plumber threads copper pipe or a copper fitting directly onto that galvanized nipple and tightens it down.

Copper and galvanized steel are dissimilar metals — they sit far apart on the electrochemical ladder that governs how metals behave when they touch each other in the presence of water. Put them in

direct contact, run water through the connection, and you have created a battery. Not a metaphorical battery. An actual electrochemical cell that generates a measurable voltage and drives a continuous corrosion reaction.

The zinc coating on the galvanized nipple is the sacrificial metal in this cell. It begins to oxidize and dissolve, atom by atom, into the water. This happens silently inside the joint. There is no visible warning — no bubbling, no discoloration, no smell. The homeowner goes on with their life. The zinc keeps corroding. When the zinc coating is finally consumed, bare carbon steel is exposed to the same electrochemical attack — and carbon steel corrodes even faster than zinc did. At some point the nipple fails. A slow seep becomes a leak. A leak becomes a flood. The water heater that should have lasted 12 to 15 years is being replaced at 7.

None of this is the heater's fault. It is the installation.

THE SCIENCE: HOW GALVANIC CORROSION ACTUALLY WORKS

To understand why this installation fails — and why the correct fix works — you need to understand galvanic corrosion at the electrochemical level. This is not complicated, but it is precise.

Three Conditions Required

Galvanic corrosion requires exactly three things to occur:

1. Two dissimilar metals
2. Direct electrical contact between them
3. An electrolyte — a fluid that conducts electricity — bridging the two metals

Tap water is an electrolyte. As ASPE's Pipeline technical journal notes, "Potable water is a weak but effective electrolyte, so failure can occur in drinking water systems."¹ When those three conditions are present, the two metals form a galvanic cell. One metal becomes the anode — it oxidizes and corrodes. The other becomes the cathode — it is protected. Which metal plays which role is determined by where each sits in the galvanic series.

The Galvanic Series: Ranking Metals by Electrochemical Potential

The galvanic series ranks metals by their electrochemical potential in a common electrolyte. Researchers measured this series specifically in tap water, giving us the most relevant data for residential plumbing.²

Metal	Potential (mV vs. Ag/AgCl)	Role in Plumbing
Copper (C11000)	+73 mV	Cathode — protected
Brass (C28000)	-67 mV	Intermediate buffer
Zinc (galvanized coating)	-927 mV	Anode — corrodes

The rule is straightforward: the more cathodic metal is protected; the more anodic metal corrodes. Copper (+73 mV) is the cathode. Zinc (-927 mV) is the anode. When copper connects directly to galvanized steel, the copper is protected — and the zinc is consumed.

The Voltage That Drives the Damage

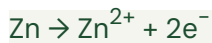
The potential difference between copper and zinc in tap water is approximately 1,000 millivolts — roughly 1 volt. Standard electrochemistry confirms this: copper has a standard electrode potential of +0.34 V and zinc has a standard electrode potential of -0.76 V, producing a theoretical cell voltage of 1.10 V.³

For context: corrosion research notes that a metal higher on the galvanic series will corrode one lower on the list when the potential difference exceeds approximately 50 mV.² The copper-to-zinc pairing in a water heater connection exceeds that threshold by a factor of 20.

What the Reaction Actually Looks Like

Here is what is happening inside that joint, in chemical terms:

At the galvanized nipple (the anode), zinc oxidizes:



Zinc atoms lose two electrons and dissolve into the water as zinc ions. Those free electrons travel through the metal from the galvanized nipple through the threaded connection into the copper pipe. At the copper surface (the cathode), those electrons participate in reduction reactions — typically reducing dissolved oxygen or hydrogen ions in the water. The water completes the ionic circuit, carrying charged particles between the two metal surfaces.

It is, in every meaningful sense, a battery. The zinc is the fuel. The water is the electrolyte. The copper pipe is the load. And the joint between them is where the damage concentrates.

"When two dissimilar metals — like a copper supply line and a steel water heater nipple — are joined together. When water (an electrolyte) flows through them, a small electrical current is created, which literally eats away at the softer metal."⁴

"When copper is connected with other metals (e.g., steel, galvanized steel) in the piping system conveying water, a galvanic cell is created, causing slow degradation of the more anodic metal."¹

Why It Goes Undetected Until Failure

The corrosion is entirely internal. It occurs at the threaded interface between the galvanized nipple and the copper fitting — inside the pipe, out of sight, under any insulation or cover plate. There is no external rust stain, no weeping, no odor. The homeowner has no indicator that the joint is failing until the zinc coating is depleted, the bare steel begins to corrode through, and water finds a path out. By then the damage is done.

WHY BRASS IS THE CORRECT SOLUTION

Brass Sits in the Middle

The fix requires inserting a brass fitting between the galvanized nipple and the copper pipe. The reason this works comes directly from the galvanic series numbers:

Copper (C11000): +73 mV

Brass (C28000): -67 mV

Zinc (galvanized coating): -927 mV

Brass sits almost exactly between copper and zinc. The potential difference between copper and brass is approximately 140 mV — compared to the 1,000 mV driving force of the copper-to-zinc pairing. That is an 86 percent reduction in galvanic driving force at the critical transition point.

Brass is not perfect — 140 mV technically still exceeds the ~50 mV threshold where minor galvanic activity begins. But it dramatically slows the electrochemical reaction at every interface in the connection. This is why brass is the industry-accepted transition metal for this application. Brass also forms a surface oxide layer over time that acts as a partial barrier, further slowing ion transfer.⁵

"Avoid connecting two dissimilar metals that are far apart on the galvanic series, because galvanic corrosion occurs when they are in direct contact or have electrical contact in the presence of an electrolyte."⁵

"Severity of corrosion increases when combining galvanized steel with metals far away from zinc in the galvanic series."⁶

Copper is about as far away from zinc as it is possible to get in common plumbing metals. Brass is not. That is the entire argument.

WHAT A CORRECT INSTALLATION LOOKS LIKE

The following describes the exact method Plumbing Detectives uses on every water heater installation in Southwest Florida. Every transition from galvanized to copper goes through brass. There is no direct copper-to-galvanized contact anywhere in the system.

Cold Side

The cold supply line feeds into the top of the tank through the cold inlet nipple, which is galvanized steel. The connection sequence:

Step 1: A brass tee is threaded directly onto the galvanized cold nipple. Brass-to-galvanized contact is the correct transition — the potential difference is manageable, and brass is the accepted buffer metal for this purpose.

Step 2: Off the top port of the brass tee, the expansion tank is installed. (See Florida code note below.)

Step 3: Out the run of the tee, a copper male adapter is used. This is a copper-to-brass connection — approximately 140 mV potential difference, roughly 86 percent less aggressive than copper-to-zinc. This is the correct and accepted transition point.

Step 4: A short piece of copper pipe connects from the copper male adapter to a new copper ball valve — the shutoff for this water heater.

Step 5: The copper ball valve connects to the copper cold supply line continuing into the house.

The galvanized nipple never sees copper. The full transition: galvanized steel → brass tee → copper male adapter → copper. Brass is the buffer.

Hot Side

The hot outlet nipple at the top of the tank is also galvanized steel. The connection sequence:

Step 1: A brass female union is threaded directly onto the galvanized hot nipple.

Brass-to-galvanized is the correct pairing for this transition.

Step 2: A copper male adapter threads into the other side of the brass union — the same brass-to-copper transition (~140 mV) used on the cold side.

Step 3: Copper pipe continues from the copper male adapter to the hot distribution system throughout the house.

The full transition: galvanized steel → brass female union → copper male adapter → copper.

The Union Detail: Built-In Serviceability

The brass female union on the hot side is not just about galvanic protection — it is professionalism built into the installation. A union allows the connection to be broken with a wrench. When this water heater reaches the end of its service life in 10 to 12 years, the next plumber can disconnect the hot side without cutting pipe. A coupling means the next service call involves cutting copper. A union means the next service call involves turning a wrench.

Unions cost two dollars more than a standard fitting. Every installer should use them on the hot side. Most don't.

The Expansion Tank: Florida Code Requirement

The expansion tank on the cold side is not optional in Florida. Under IRC P2903.4, an expansion tank is required on all closed plumbing systems — systems where a backflow prevention device prevents water from returning to the municipal supply line. Florida requires backflow prevention devices on virtually all homes served by public water, making virtually every home on public water in Florida a closed system. In a closed system, thermal expansion of heated water has nowhere to go. Without an expansion tank, that pressure builds against every fitting, valve, and appliance in the system.

The expansion tank must be installed on the cold supply side — which is why it comes off the top of the cold-side brass tee in this installation.

HOW TO CHECK YOUR OWN WATER HEATER

You do not need to be a plumber to identify a potentially problematic installation. Here is what to look for:

Step 1: Locate the connections at the top of your water heater.

The two pipes coming out of the top are the cold inlet (usually on the right, often with a blue label or blue shutoff handle) and the hot outlet (usually on the left).

Step 2: Identify the nipples at the top of the tank.

If the nipples — the short pipe sections threaded directly into the tank — are silver-colored with a somewhat rough, matte surface, they are likely galvanized steel. Many modern heaters use dielectric nipples with a plastic liner. If you are unsure, call a plumber.

Step 3: Identify the first fitting above each nipple.

If the first fitting above the galvanized nipple is copper — a copper male adapter, copper elbow, or copper pipe threaded directly in — that is the problem. There is no brass buffer. Galvanic corrosion is active.

Step 4: Look for early warning signs.

Examine the connection area for: white or greenish deposits around the threads (zinc or copper corrosion byproducts), rust staining running down from the connection, or any dampness on the top of the tank near the connections. If you see any of these, have a licensed plumber evaluate the installation.

Step 5: Check for an expansion tank.

If you are on public water in Florida, you should have a small cylindrical tank — typically 2 gallons — connected to the cold supply line near the water heater. If you do not see one, your system may not be code-compliant and thermal expansion is stressing your connections.

If your installation has any of these issues, the correct repair is not to seal around the threads with tape or pipe dope. The correct repair is to remove the copper fittings and reinstall with proper brass transition fittings. In most cases this is a 2- to 3-hour job that costs far less than the water damage from a failed connection — or the cost of replacing a heater 5 years early because the nipple corroded through.

ABOUT PLUMBING DETECTIVES LLC

Plumbing Detectives LLC is a licensed plumbing contractor (CFC1434139) serving Venice, Sarasota, Englewood, North Port, Nokomis, and the surrounding Southwest Florida area. Owner Mike Reynolds and his team specialize in water heater installation and replacement, whole-home repiping, and plumbing diagnostics for both residential and light commercial properties. Every installation is done to code and built to last.

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SOURCES

1. ASPE Pipeline. "Plumbing Design for Success: Understanding and Preventing Copper Corrosion." <https://aspe.org/pipeline/plumbing-design-for-success-understanding-and-preventing-copper-corrosion/>
2. Matsukawa et al. (2011). "Galvanic Series of Metals Conventionally Used in Tap Water." Corrosion, v. 67. Via http://www.sedimentaryores.net/Pipe%20Scales/Galvanic/Galvanic_series.html
3. Princeton University MAE 324 Electrochemistry Lab. Standard electrode potentials. <https://www.princeton.edu/~maelabs/mae324/12/galvaniccell.htm>
4. Allied Plumbing & Heating. "Why Water Heater Connections Fail." <https://www.alliedpnh.com/blogs/why-water-heater-connections-fail>
5. Clear Water Industries. "Galvanic Corrosion in Water Systems: Causes & Prevention." <https://clearwatershelton.com/galvanic-corrosion/>
6. American Galvanizers Association. Galvanic corrosion and zinc. <https://galvanizeit.org/design-and-fabrication/design-considerations/dissimilar-metals-in-contact>